

## Chapter 16 Thermal Energy And Heat Section 2 Thermodynamics Answer Key

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### Chapter 16 Thermal Energy And

Chapter 16 Thermal Energy and Heat. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Sokollm (In order of which they appear) Key Concepts: Terms in this set (20) Heat. the transfer of thermal energy from one object to another because of a difference in temperature. Temperature. a measure of how hot or cold an ...

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## **Chapter 16 Thermal Energy and Heat**

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It states that thermal energy can flow from colder objects to hotter objects only if work is done on the system. Third law of thermodynamics It states that absolute zero cannot be reached.

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is a measure of how hot or cold an object is compared to a ref.... Increase in volume of material when its temperature increases. Conduction. is the transfer of thermal energy through touching with no ove.... Heat Engine. is any device that converts heat into work. 36 Terms. etakp. Chapter 16 Thermal Energy and Heat.

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## **Chapter 16 - Thermal Energy**

16.1 Thermal Energy & Matter. Work and Heat. Heat -the transfer of thermal energy from one object to another because of a temperature difference Heat flows from higher temps to lower temps. Temperature is related to the kinetic energy of the particles: particles move around as they heat

## **Chapter 16**

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Chapter 16: Thermal Energy and Heat. Tools. Copy this to my account; E-mail to a friend; Find other activities; Start over; Help; A B; heat: the transfer of thermal energy from one object to another because of a difference in temperature: temperature: a measure of how hot or cold an object is compared to a reference point:

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Chapter 16 Thermal Energy and Heat Section 161 Thermal Energy and Matter (pages 474-478) This section defines heat and describes how work, temperature, and thermal energy are related to heat Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained Reading Strategy (page 474)

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Chapter 16 Thermal Energy and Heat Calculating with Specific Heat How much heat is required to raise the temperature of a gold earring from 25.00C to 30.00C? The earring weighs 25 grams, and the specific heat of gold is 0.128 J/ge0C. 1. 2. 3. Read and Understand What information are you given? Specific heat =  $c = 0.128 \text{ J/g} \cdot \text{OC}$  Mass =  $m = 25.0 \text{ grams}$

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Heat. the transfer of thermal energy from one object to another as the result of a difference in temperature. True. T/F: On the Celsius Scale, the reference points for temperature are the freezing and boiling points of water. thermal energy ...

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