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question_answer Which of the following
statement is false [BHU 1994] . A) The
product of pressure and volume of fixed
amount of a gas is independent of
temperature . B) Molecules of different gases

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have the same K.E. at a given temperature

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gases and Molecular collisions ... Kinetic energy of the gas becomes zero but the molecular motion does not become zero .
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_____ is water in its gaseous form. ... The conversion of solid into gas without changing into liquid is known as _____ A) Heating done clear. B) Evaporation done clear. C) Melting ...

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and Gas . question_answer Which of the
following has fixed volume ...

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What is the composition of natural Gas:
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and Gas . question_answer. In the rainy
seasons, there is a lot of humidity. ...

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(i) It is an inflammable gas (ii) It is the

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Tightest gas. (iii) It is not easily liquefied. (iv) It is a strong oxidizing agent. (e) For the reaction $\text{PbO} + \text{H}_2 \rightarrow \text{Pb} + \text{H}_2\text{O}$, which of the following statements is wrong? i) H_2 is the reducing agent. (ii) PbO is the oxidizing agent. (iii) PbO is oxidized to Pb (iv) H_2 is oxidized to H_2O .

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the product formation after heating the
Calcium carbonate.

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